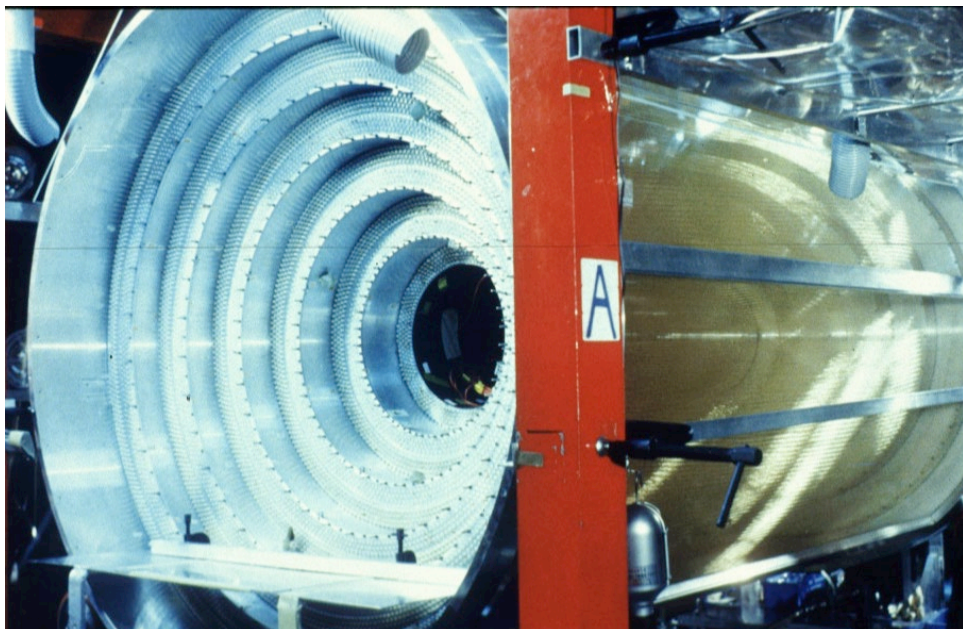
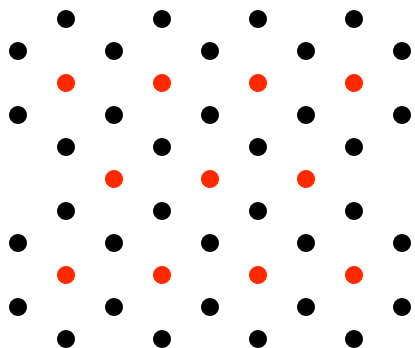


Drift Chambers

AMY experiment at e^+e^- TRISTAN collider



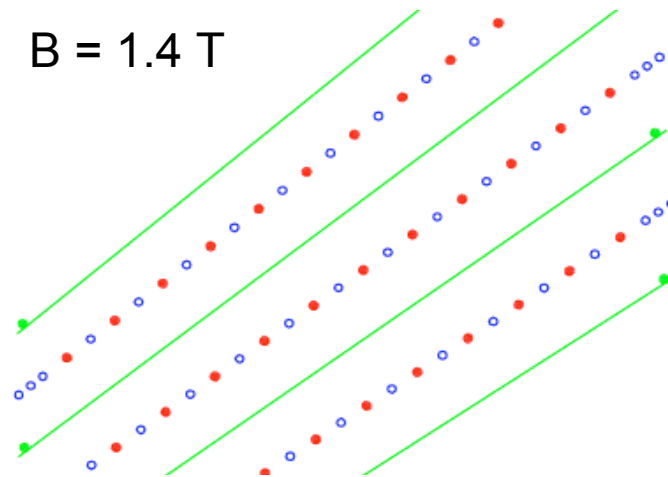
$B = 3 \text{ T}$



CDF experiment

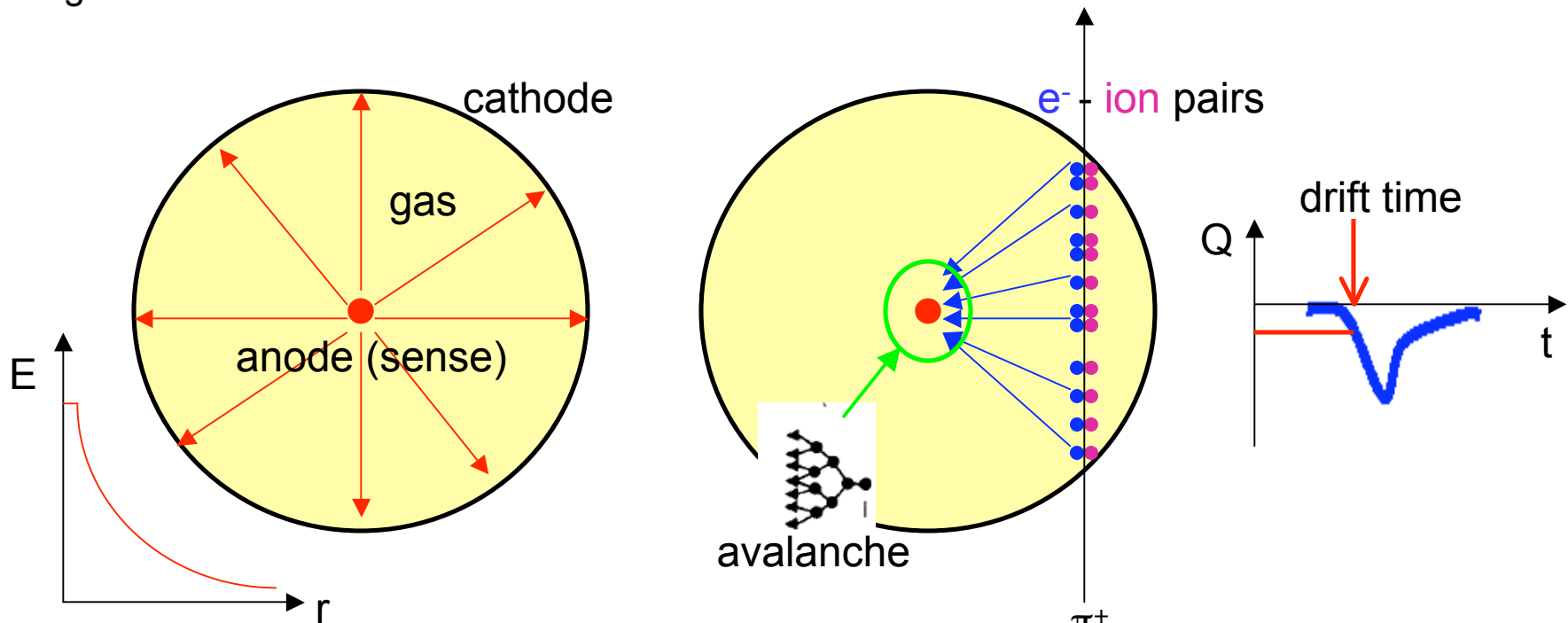


$B = 1.4 \text{ T}$



E-field, Drift-Time Relation

- A charged particle enters gas in E-field, ionizes gas, produces e⁻-ion pairs
 - ~300 μm / pair
 - Ionization E ~30 eV / pair
- Primary ionization electrons drift toward anode (sense) wire (low E field region)
- Avalanche multiplication of charges by electron-atom collision in high E field region - within a few radii of the wire.
- Signal induced via motion of charges.
 - Measure “drift time”, Δt_{drift} , (first arrival time) of electrons at sense wire relative to a time t_0 (e.g. collision time)
 - Locate the position of “first” electrons
 - $D = \int_{t_0}^{t_0 + \Delta t} v(t) dt$
 - $D = v \Delta t_{\text{drift}}$ if the drift velocity is independent of E field.

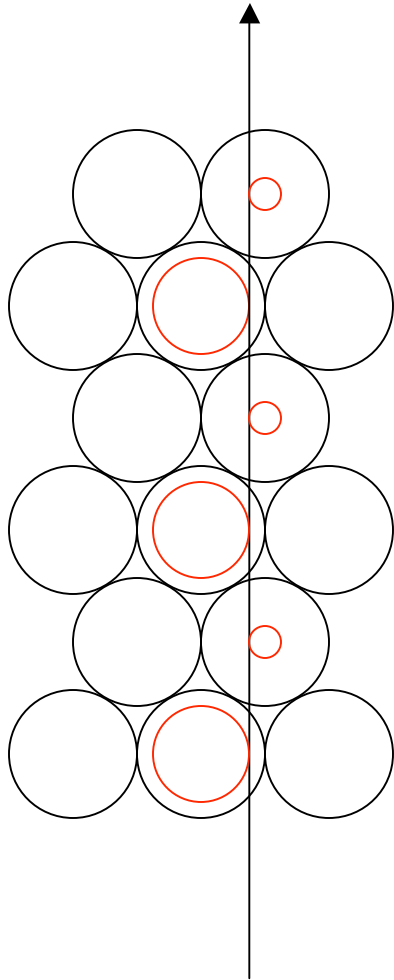


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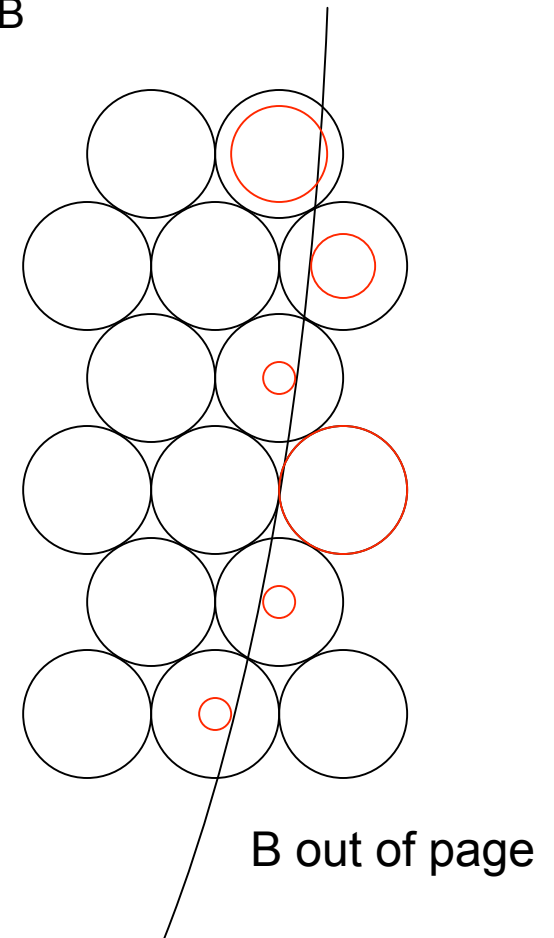
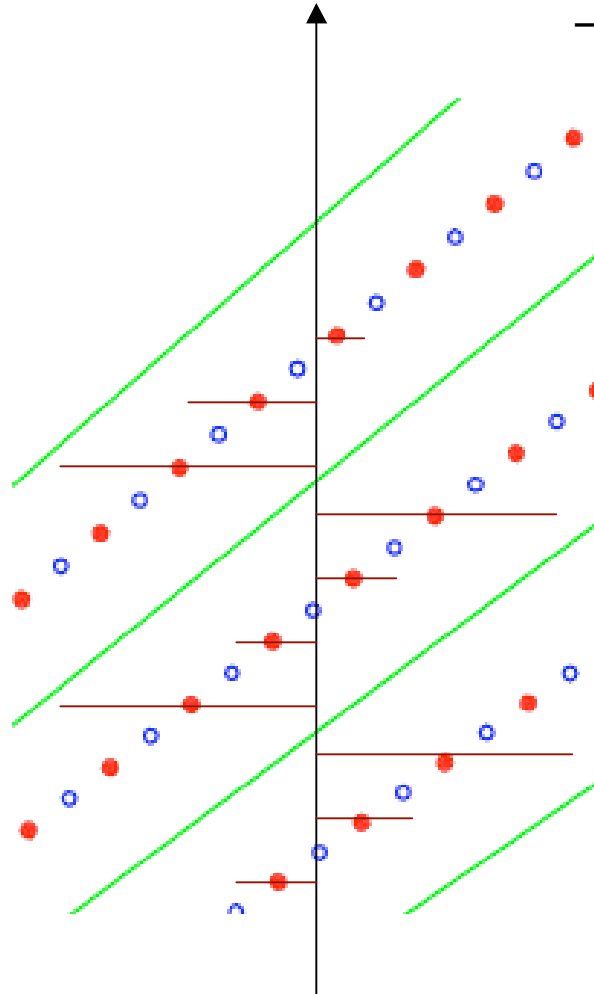
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Left-Right Ambiguity, B field

- Left-right ambiguity
- solved by staggering



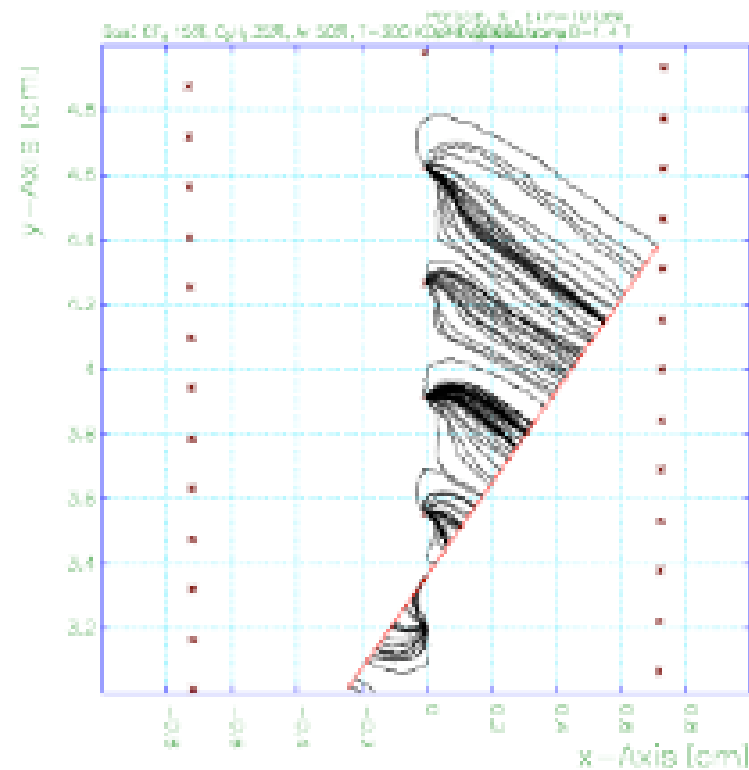
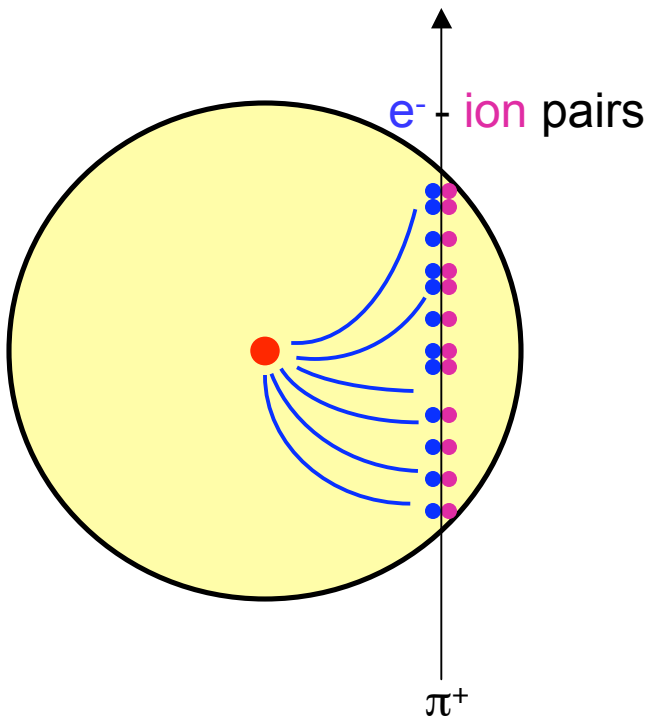
- Apply Magnetic field to measure momentum
 - Measure curvature, $C = 1 / R$
 - $P = 0.3 RB$



B-fields, Lorentz Angle

- Magnetic fields complicate the drift-time relation
- Introduce the drift angle (Lorentz angle).

- COT
 - Argon: Ethane = 50:50
 - Lorentz angle of $\sim 30^\circ$
 - Tilt cells



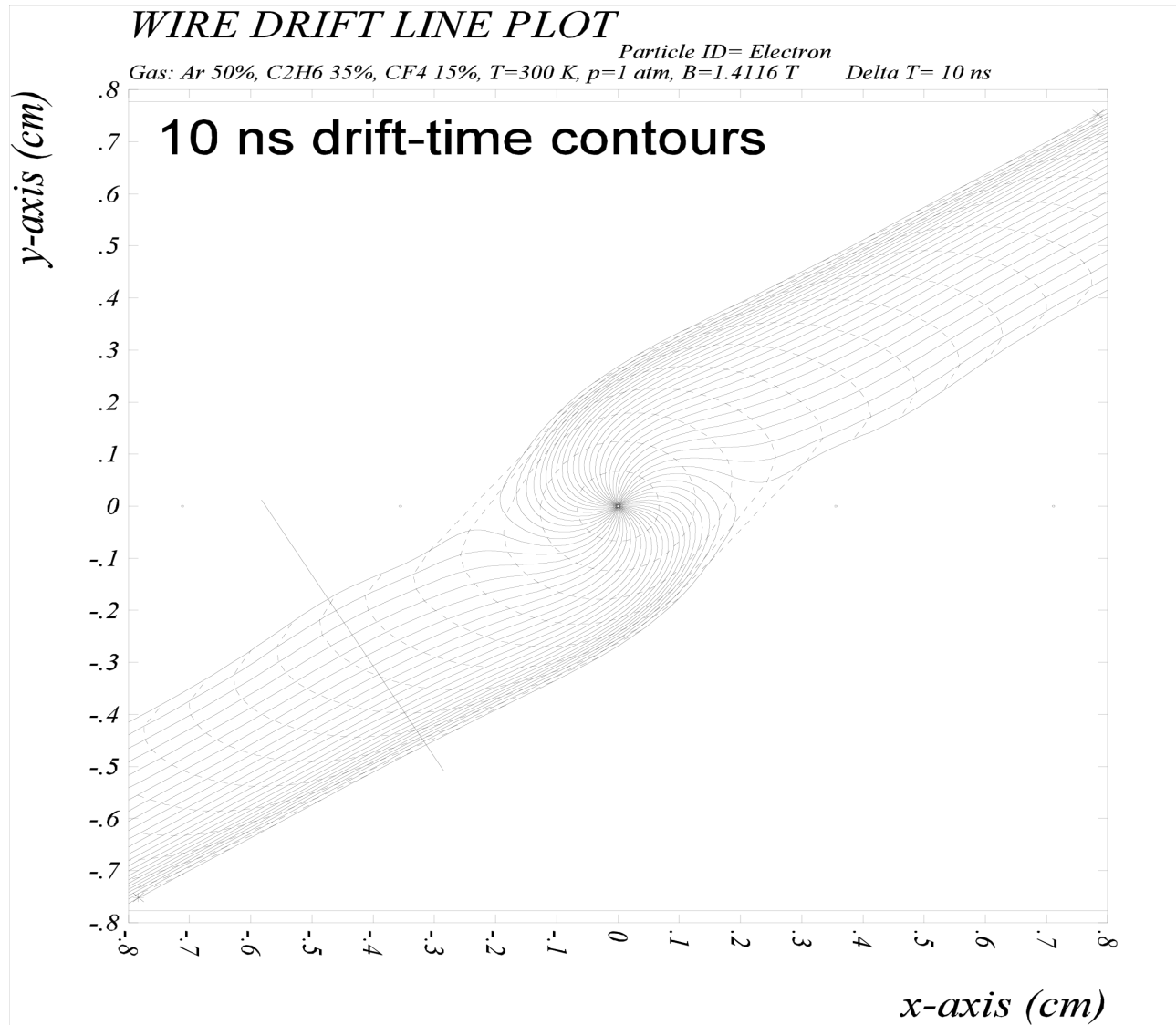
Choice of Gas (desired property)

- Drift velocity independent of E field
 - Good linear distance-time correlations.
 - $v \sim 50 \mu\text{m} / \text{ns}$
- Low working voltage
- High gain operation
- Small diffusion coefficient for electrons traveling through the gas.
 - Amount of diffusion that occurs limits the spatial resolution obtainable
- Argon is widely used as a drift chamber gas. The argon may be mixed with gases consisting of heavy organic molecules (“quencher”).
- Quenchers - heavy organic molecules that have many degrees of freedom and can efficiently absorb energy from the gas.
 - The effective temperature of the electron is reduced.
 - The drift velocity is increased.
 - The diffusion is decreased.
 - Carbon dioxide, isobutane, ...
- Avoid elements with strong affinity of electrons.
 - Otherwise the detection efficiency will depend on the drift distance.
 - Avoid O_2 , H_2O , F, Cl

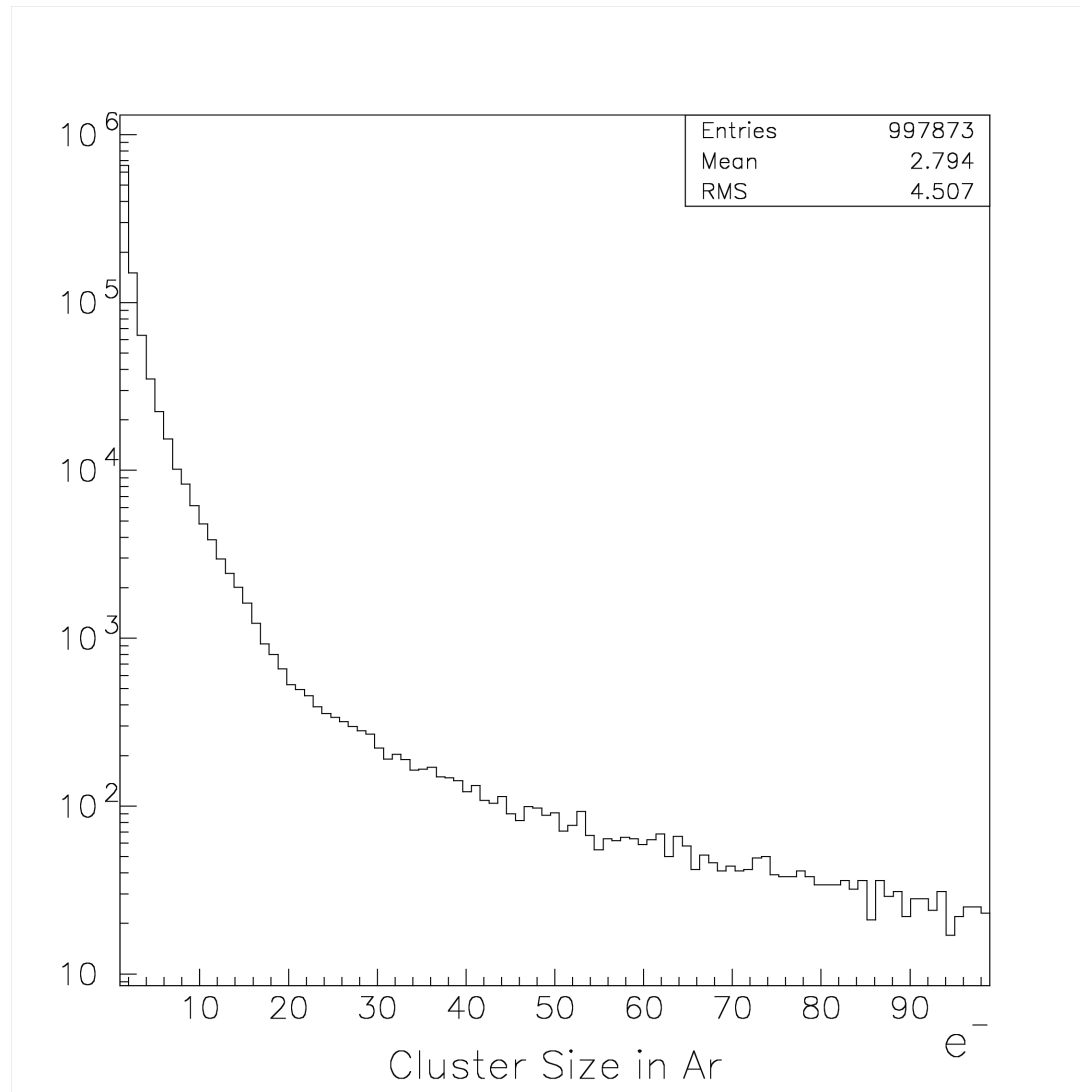
Signal formation, dE/dx

- What happens
 - Charged track traverses the cell and deposits a number of clusters-Poisson distributed
 - Clusters are distributed along the track trajectory uniformly
 - The number of electrons per cluster follows a particular distribution, peaked at about one, mean about 3 in Argon, but with long tail-Landau like tail
 - Drift electrons in along field lines-drift time depends on gas, field, trajectory to the wire
 - Avalanche takes place near wire-Polya distribution
 - Signal forms from electron pulse (10%) and ion drift to cathode-preamplifier cuts this off-pulse shape formed

Drift time contours (isochrones)



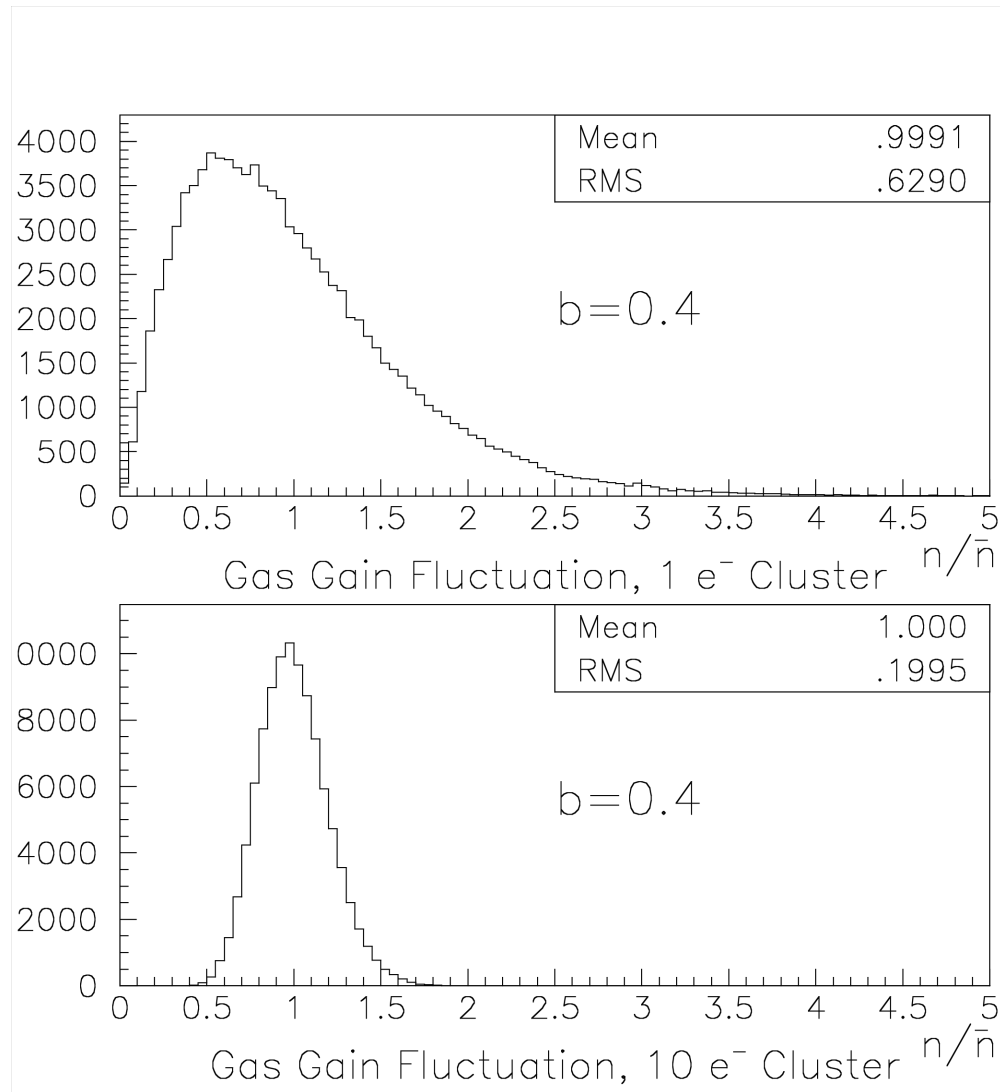
Electrons per Cluster



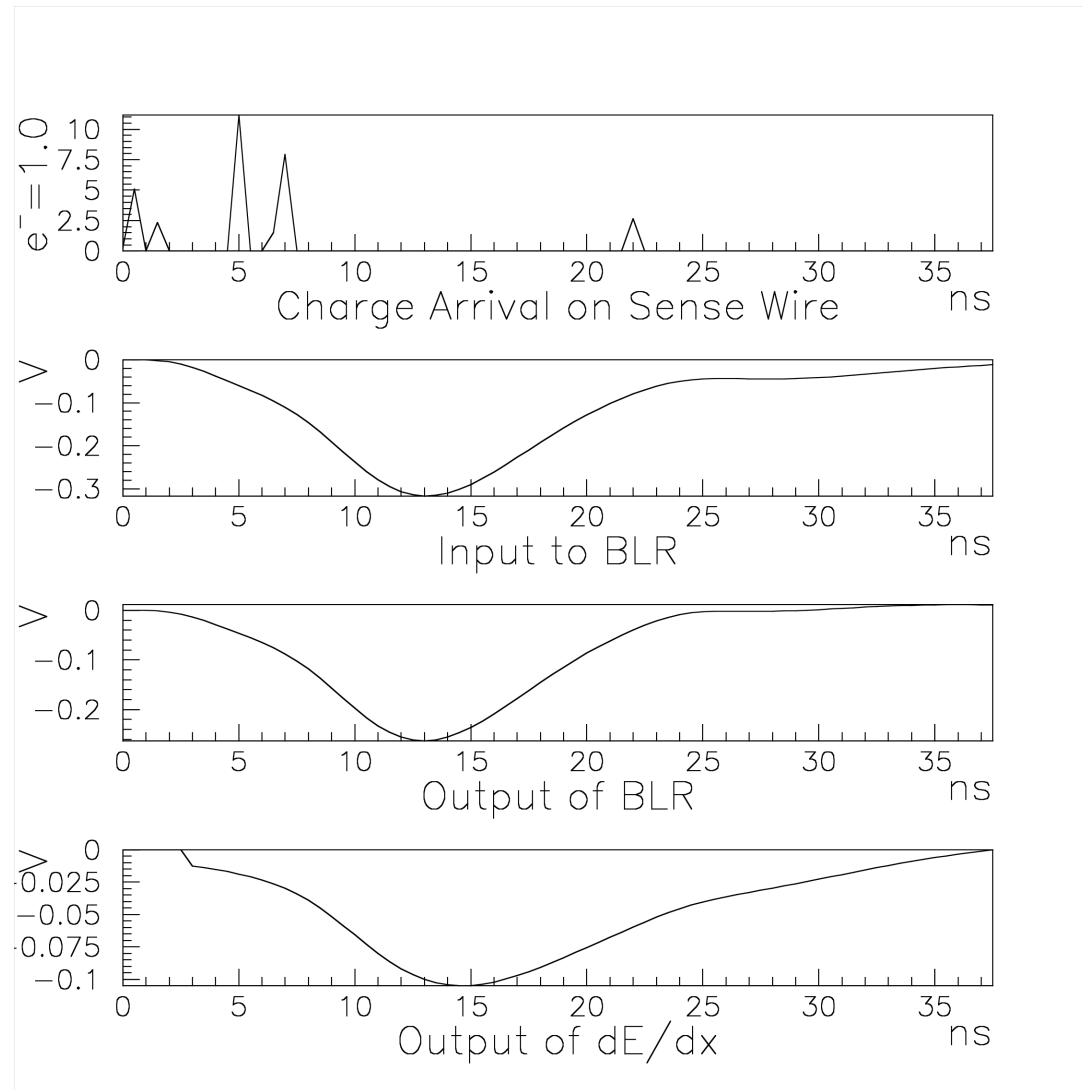
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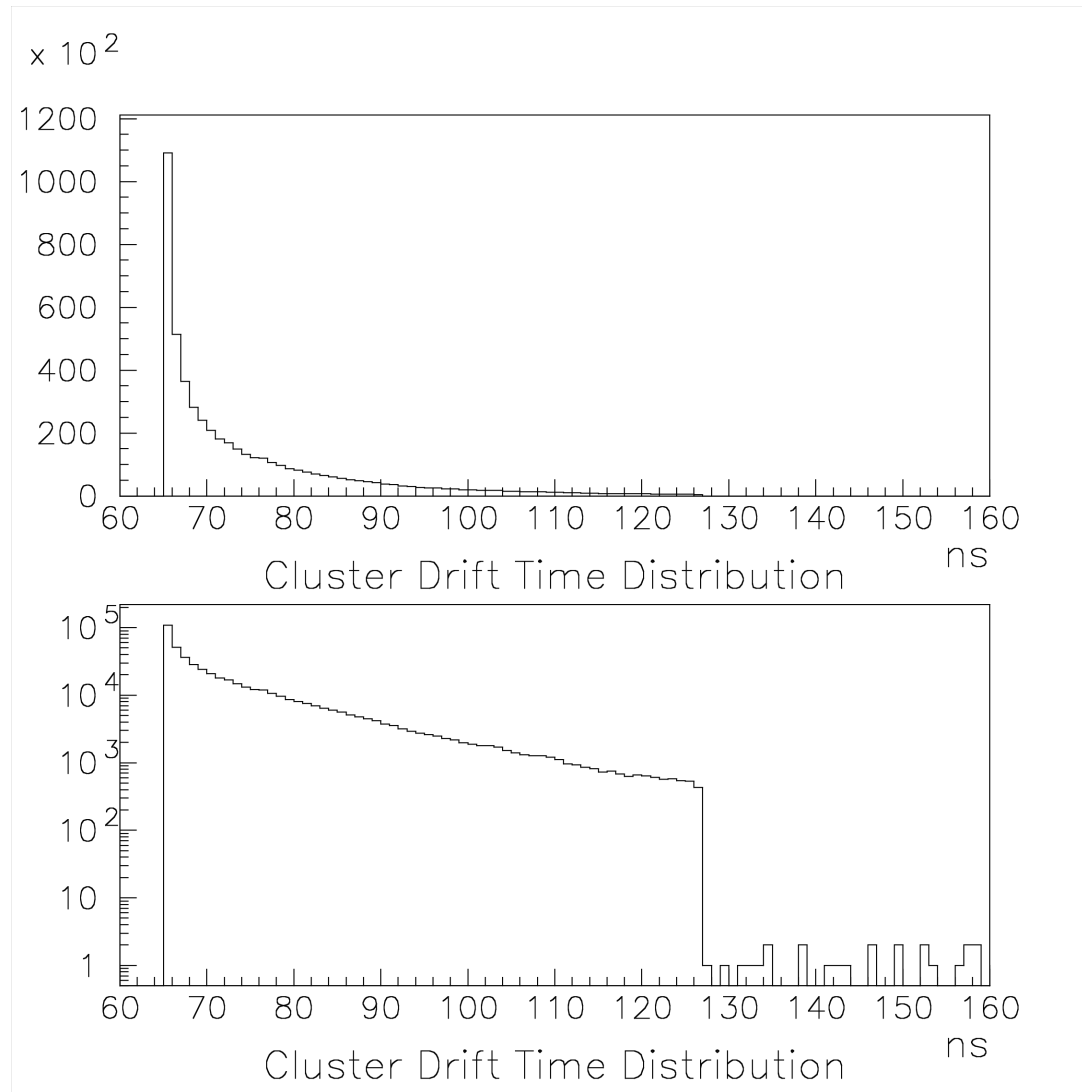
Polya Distribution



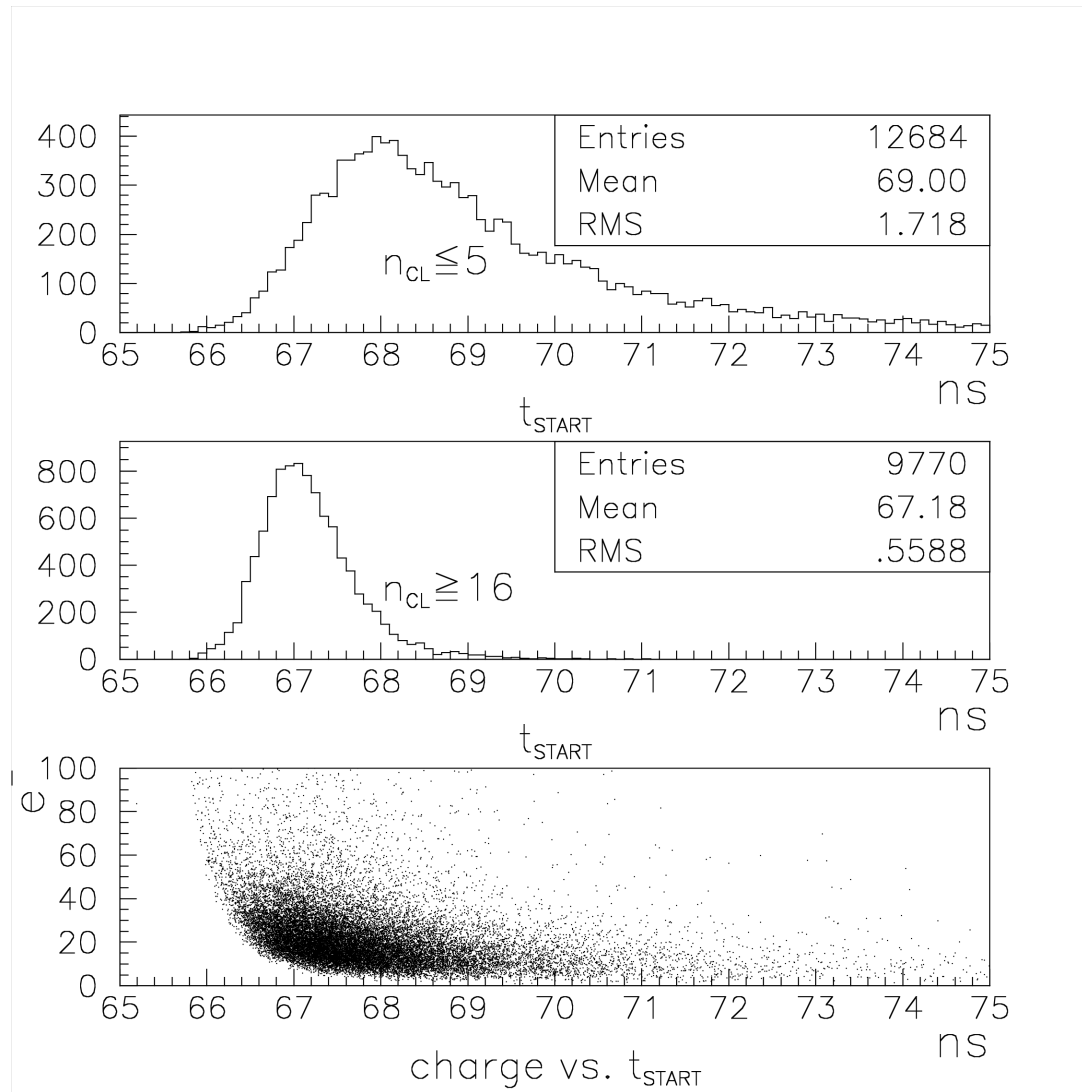
Signal Formation



Cluster Time Distribution



Charge versus T_{start}



Charge versus Width

